## WHAT IS CLAIMED IS:

1	1.	An alkaline battery comprising:	
2	a cathode comprising an additive, the additive including a barium salt and an		
3	electrically conductive material, and a manganese dioxide;		
4	an anode comprising zinc;		
5	a separator between the cathode and the anode; and		
6	an alkaline electrolyte in contact with the cathode and the anode.		
1	2.	The battery of claim 1, wherein the electrically conductive material is coated	
2	on a surface of the barium salt.		
1	3.	The battery of claim 1, wherein the electrically conductive material includes a	
2	metal oxide.		
1	4.	The battery of claim 2, wherein the electrically conductive material includes a	
2	metal oxide.		
1	5.	The battery of claim 3, wherein the metal oxide is a tin oxide.	
1	6.	The battery of claim 4, wherein the metal oxide is a tin oxide.	
1	7.	The battery of claim 1, wherein the barium salt includes barium sulfate,	
2	barium hydroxide, barium carbonate, or barium oxide.		
1	8.	The battery of claim 1, wherein the manganese dioxide is an electrolytic	
2	manganese dioxide.		
1	9.	The battery of claim 1, wherein the service life of the battery in an intermittent	
2	discharge tes	t is at least 2% longer than the service life of a battery lacking the particle.	
1	10.	The battery of claim 1, wherein the service life of the battery in an intermittent	
2	discharge tes	discharge test is at least 3% longer than the service life of a battery lacking the particle.	
1	11.	An alkaline battery comprising:	
2	a cathode comprising an additive, the additive including a barium salt and a coating		

on a surface of the barium salt, and a manganese dioxide; 3 an anode comprising zinc; 4 a separator between the cathode and the anode; and 5 an alkaline electrolyte in contact with the cathode and the anode. 6 12. The battery of claim 11, wherein the coating is electrically conductive. 1 13. The battery of claim 12, wherein the coating includes a metal oxide. 1 14. The battery of claim 13, wherein the metal oxide is a tin oxide. 1 15. The battery of claim 11, wherein the barium salt includes barium sulfate, 1 barium hydroxide, barium carbonate, or barium oxide. 2 The battery of claim 11, wherein the manganese dioxide is an electrolytic 16. 1 manganese dioxide. 2 17. The battery of claim 11, wherein the service life of the battery in an 1 intermittent discharge test is at least 2% longer than the service life of a battery lacking the 2 particle. 3 The battery of claim 11, wherein the service life of the battery in an 18. 1 intermittent discharge test is at least 3% longer than the service life of a battery lacking the 2 particle. 3 A method of manufacturing an alkaline battery comprising forming a cathode 19. 1 including a cathode active material including a manganese dioxide, and an additive including 2 a barium salt and an electrically conductive material. 3 20. The method of claim 19, wherein the electrically conductive material is coated 1

The method of claim 19, wherein the electrically conductive material includes

on a surface of the barium salt.

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a metal oxide.

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22. The method of claim 20, wherein the electrically conductive material includes 1 a metal oxide. 2 23. The method of claim 21, wherein the metal oxide is a tin oxide. 1 24. The method of claim 22, wherein the metal oxide is a tin oxide. 1 25. The method of claim 19, wherein the barium salt includes barium sulfate, 1 barium hydroxide, barium carbonate, or barium oxide. 2 26. The method of claim 19, wherein the manganese dioxide is an electrolytic 1 manganese dioxide. 2 27. The method of claim 19, further comprising assembling the cathode with an 1 anode, a separator, and an electrolyte in a housing. 2 A method of making an alkaline battery comprising: 28. 1 combining an additive, the additive including a barium salt and a coating on a surface 2 of the barium salt, with a cathode active material including a manganese dioxide. 3 29. The method of claim 28, wherein the coating is electrically conductive. 1 The method of claim 29, wherein the coating includes a metal oxide. 30. 1 The method of claim 30, wherein the metal oxide is a tin oxide. 31. 1 The method of claim 28, wherein the barium salt includes barium sulfate, 32. 1 barium hydroxide, barium carbonate, or barium oxide. 2 The method of claim 28, wherein the manganese dioxide is an electrolytic 33. 1 manganese dioxide. 2 34. The method of claim 28, further comprising assembling the cathode with an 1 anode, a separator, and an electrolyte in a housing. 2

1	35.	A method of increasing the service life of an alkaline battery comprising	
2	adding an additive, the additive including a barium salt and a coating on a surface of the		
3	barium salt, to	a cathode active material including a manganese dioxide.	
1	36.	The method of claim 35, wherein the coating is electrically conductive.	
1	37.	The method of claim 35, wherein the coating includes a metal oxide.	
1	38.	The method of claim 37, wherein the metal oxide is a tin oxide.	
	20	The weether defending 25 whomein the horizon gold includes herizon sulfate.	

- 1 39. The method of claim 35, wherein the barium salt includes barium sulfate, 2 barium hydroxide, or barium oxide.
  - 40. The method of claim 35, wherein the manganese dioxide is an electrolytic manganese dioxide.

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- 41. The method of claim 35, wherein the service life of the battery in an intermittent discharge test is at least 2% longer than the service life of a battery lacking the particle.
- 42. The method of claim 35, wherein the service life of the battery in an intermittent discharge test is at least 3% longer than the service life of a battery lacking the particle.